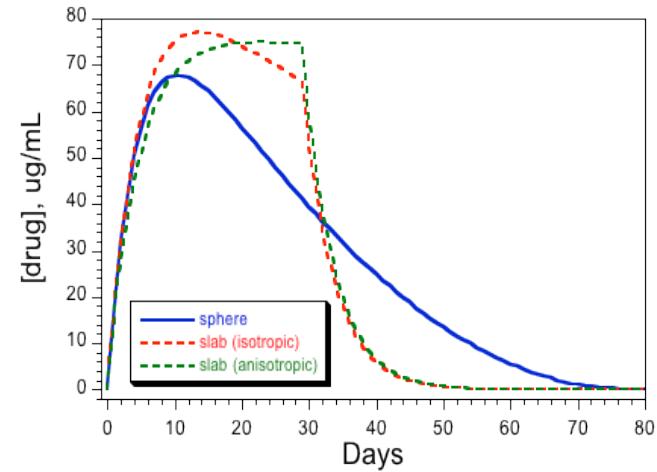
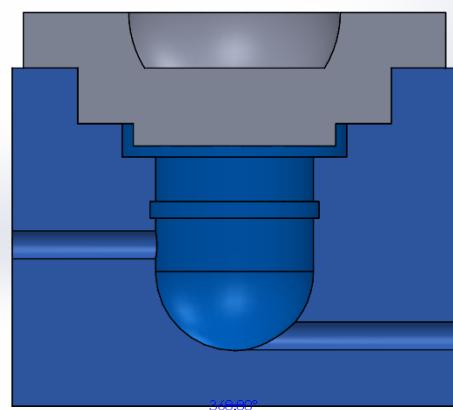
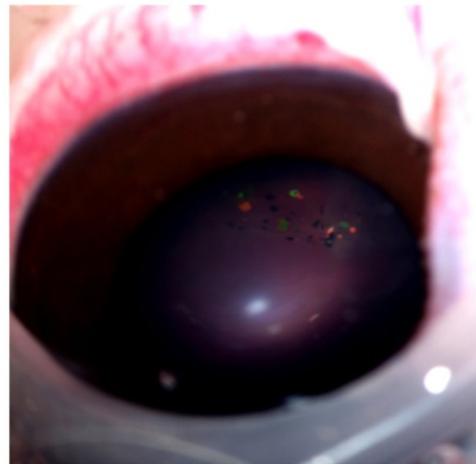


# *In vitro drug release testing of ophthalmic suspensions*

*Demonstrating Equivalence of Generic  
Complex Drug Substances and Formulations*

FDA White Oak  
6 October, 2017



FDA contract 1U01FD005173-01



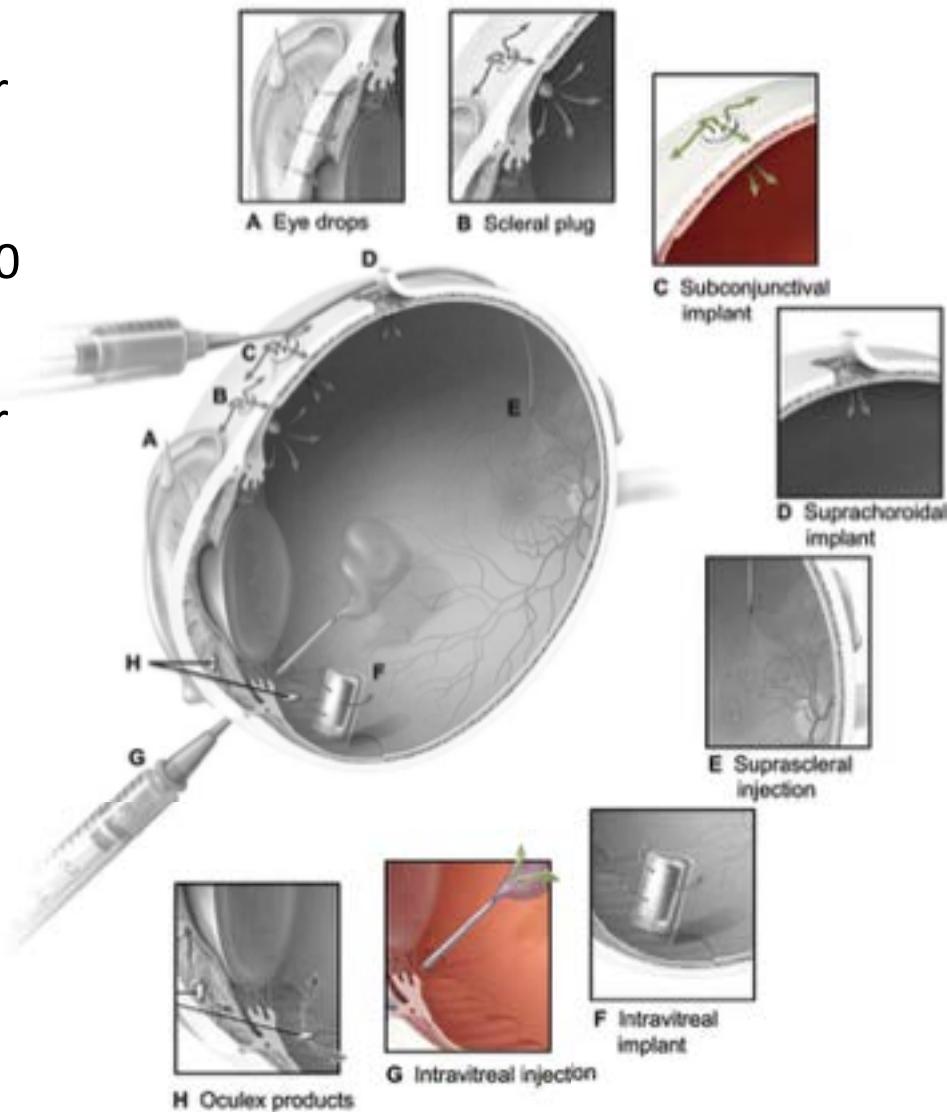
# *Long-acting intraocular drug delivery*

- 9 million Americans over 40 suffer from Age-related Macular Degeneration (500,000 new cases of “wet” AMD per year, at \$10,000 / case / year)\*
- 4 million Americans over 40 suffer from Diabetic Retinopathy\*
- Over \$5 billion spent in 2003 on drugs to treat vision disorders\*\*

## Sources:

\* Archives of Ophthalmology,  
vol. 122, Apr 2004

\*\* National Eye Institute,  
2003



# Generic nanomaterials



U.S. Department of Health and Human Services



**U.S. Food and Drug Administration**  
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## **Issues:**

*W. Jiang, R. Lionberger, L.X. Yu (FDA Office of Generic Drugs)*

*"In vitro and in vivo characterizations of PEGylated liposomal doxorubicin," Bioanalysis, 2011, 3, 333.*

*"One challenge in developing a nanoparticle drug-delivery system is understanding the critical physicochemical properties that may impact its in vivo performance and establishing analytical techniques that can adequately characterize in vitro and in vivo properties."*

- *composition*
- *state of encapsulated drug*
- *internal environment*
- *size distribution*
- *surface chemistry*
- *electrical surface potential or charge*
- *in vitro leakage*
- *in vivo stability*
- *plasma pharmacokinetics*

# Nanomaterials can be very sensitive to variations in manufacture

Brinker, C.J., et al, "Processing Pathway Dependence of Amorphous Silica Nanoparticle Toxicity: Colloidal vs Pyrolytic" *J. Am. Chem. Soc.* 2012, 134, 15790.

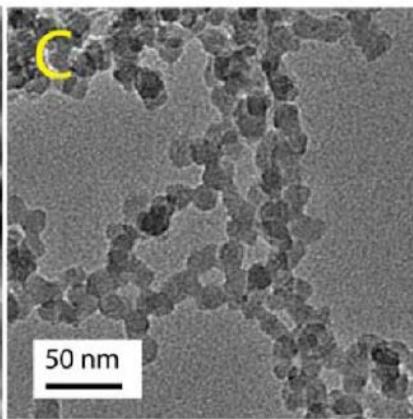
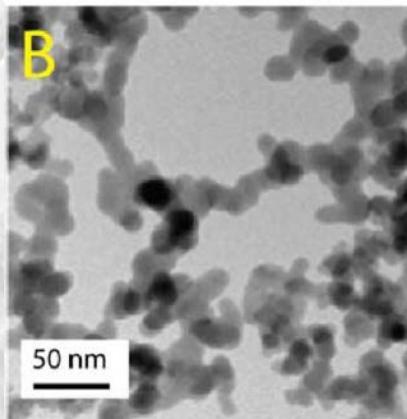
**J|A|C|S** *J. Am. Chem. Soc.* 2012, 134 (38), 15790-15804 Article pubs.acs.org/JACS

**Processing Pathway Dependence of Amorphous Silica Nanoparticle Toxicity: Colloidal vs Pyrolytic**

Haiyuan Zhang,<sup>†</sup> Darren R. Dunphy,<sup>‡</sup> Xingmao Jiang,<sup>‡,§</sup> Huan Meng,<sup>||</sup> Bingbing Sun,<sup>†</sup> Derrick Tarn,<sup>‡</sup> Min Xue,<sup>†</sup> Xiang Wang,<sup>†</sup> Sijie Lin,<sup>†</sup> Zhaoxia Ji,<sup>†</sup> Rubin Li,<sup>†</sup> Fred L. Garcia,<sup>‡</sup> Jing Yang,<sup>||</sup> Martin L. Kirk,<sup>||</sup> Tian Xia,<sup>||</sup> Jeffrey I. Zink,<sup>‡</sup> Andre Nel,<sup>†,||</sup> and C. Jeffrey Brinker<sup>‡,§,||</sup>

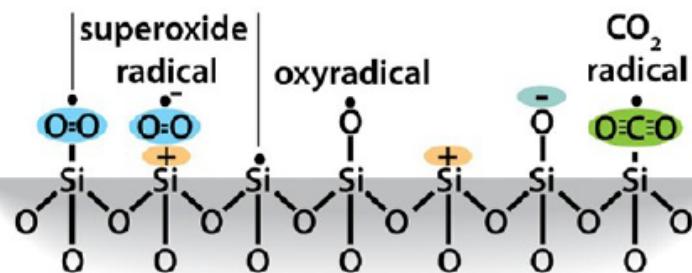
*"...we discovered for fumed silica an important toxicity relationship...whereas colloidal silicas were essentially nontoxic...not all amorphous silicas are created equal"*

**ABSTRACT:** We have developed structure/toxicity relationships for amorphous silica nanoparticles (NPs) synthesized through low-temperature colloidal (e.g., Stöber silica) or high-temperature pyrolysis (e.g., fumed silica) routes. Through combined spectroscopic and physical analyses, we have determined the state of aggregation, hydroxyl concentration, relative proportion of strained and unstrained siloxane rings, and potential to generate hydroxyl radicals for Stöber and fumed silica NPs with comparable primary particle sizes (16 nm in diameter). On the basis of erythrocyte hemolytic assays and assessment of the viability and ATP levels in epithelial and macrophage cells, we discovered for fumed silica an important toxicity relationship to postsynthesis thermal annealing or environmental exposure, whereas colloidal silicas were essentially nontoxic under identical treatment conditions. Specifically, we find for fumed silica a positive correlation of toxicity with hydroxyl concentration and its potential to generate reactive oxygen species (ROS) and cause red blood cell hemolysis. We propose fumed silica toxicity stems from its intrinsic population of strained three-membered rings (3MRs) along with its chainlike aggregation and hydroxyl content. Hydrogen-bonding and electrostatic interactions of the silanol surfaces of fumed silica aggregates with the extracellular plasma membrane cause membrane perturbations sensed by the Nalp3 inflammasome, whose subsequent activation leads to secretion of the cytokine IL-1 $\beta$ . Hydroxyl radicals generated by the strained 3MRs in fumed silica, but largely absent in colloidal silicas, may contribute to the inflammatory activation. Formation of colloidal silica into aggregates mimicking those of fumed silica had no effect on cell viability or hemolysis. This study emphasizes that not all amorphous silicas are created equal and that the unusual toxicity of fumed silica compared to that of colloidal silica derives from its framework and surface chemistry along with its fused chainlike morphology established by high-temperature synthesis ( $>1300$  °C) and rapid thermal quenching.

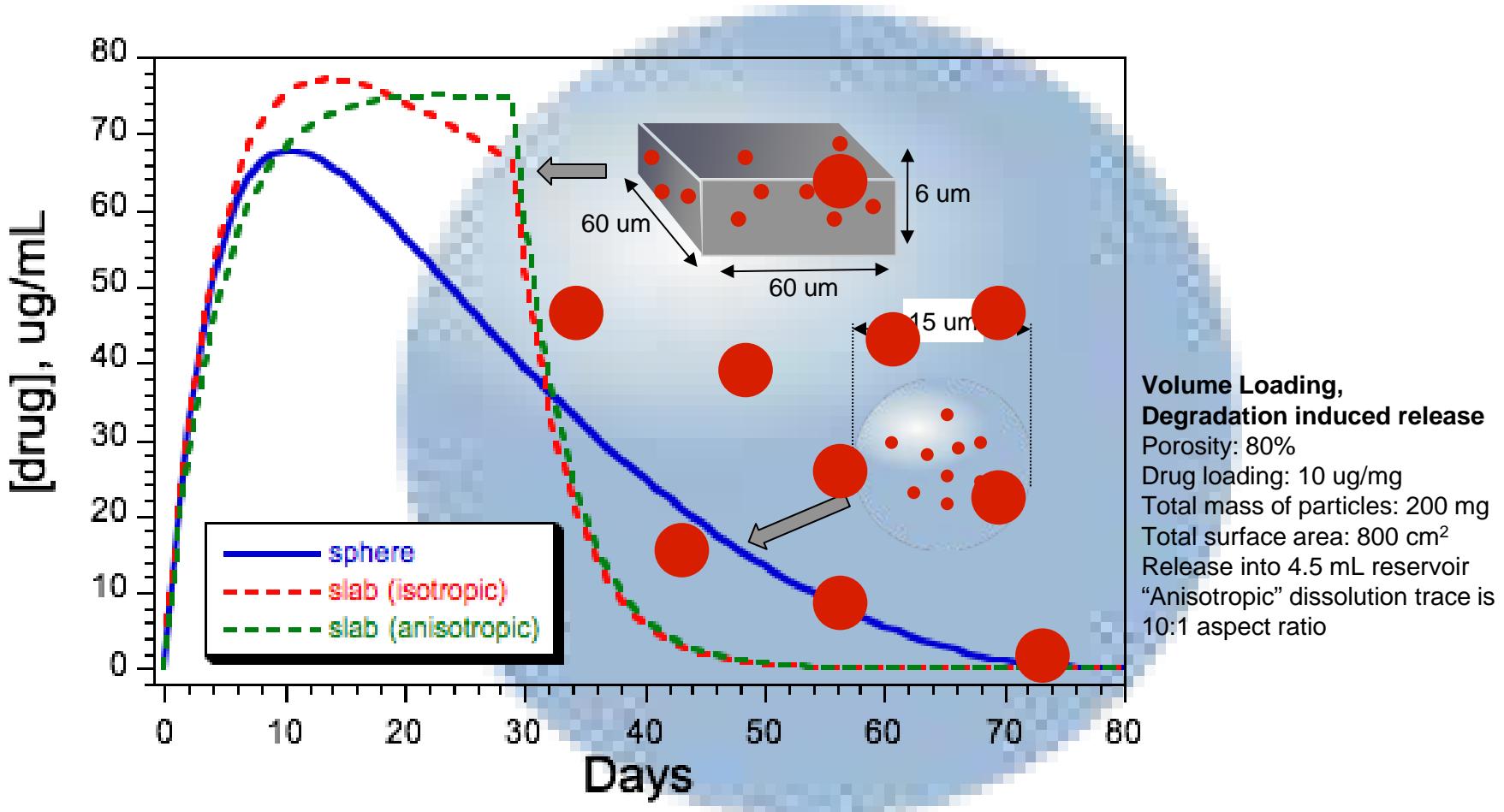


fumed silica

colloidal silica



# Particle shape determines release profile



- Tabular particles dissolve more uniformly than spherical
- Anisotropic (1-d) dissolution flattens  $\frac{\text{rate of dissolution}}{\text{dissolution front}} = k \frac{S}{V} (C - C_{\text{sat}})$  release profile

# Drug delivery systems for the eye

Name	Type/Composition	Drug	Disease	Treatment period
Vitrasert	Implant, Polyvinyl alcohol, ethylene vinyl acetate	Ganciclovir	Virus (herpes, cytomegalovirus (CMV), etc)	5-8 months
Retisert	Implant, silicone, polyvinyl alcohol, cellulose	Fluocinolone Acetonide	Uveitis	2.5 years
Ozurdex	Implant (rod), PLGA	Dexamethasone	Macular degeneration and edema, uveitis, vein occlusion	3-6 months
Iluvien	Implant (rod), polyimide, polyvinyl alcohol	fluocinolone acetonide	Diabetic macular edema (DME)	3 years
Triesence	Microsphere suspension, Tween80	Triamcinolone Acetonide	ocular inflammation	3 months

# Need for *in vitro* simulator

- Predict *in vivo* performance to optimize design of DDS
- Reduce number of animal experiments during development
- Differentiate between similar formulations for generic drug validation (FDA)

# Goals

**Design and validate a vitreous model for dissolution of particles and implants**

Flow cell design

Test different media to mimic vitreous

Conduct *in vitro* release studies

**Benchmark results against literature and *in vivo* data**

Perform *in vivo* release studies

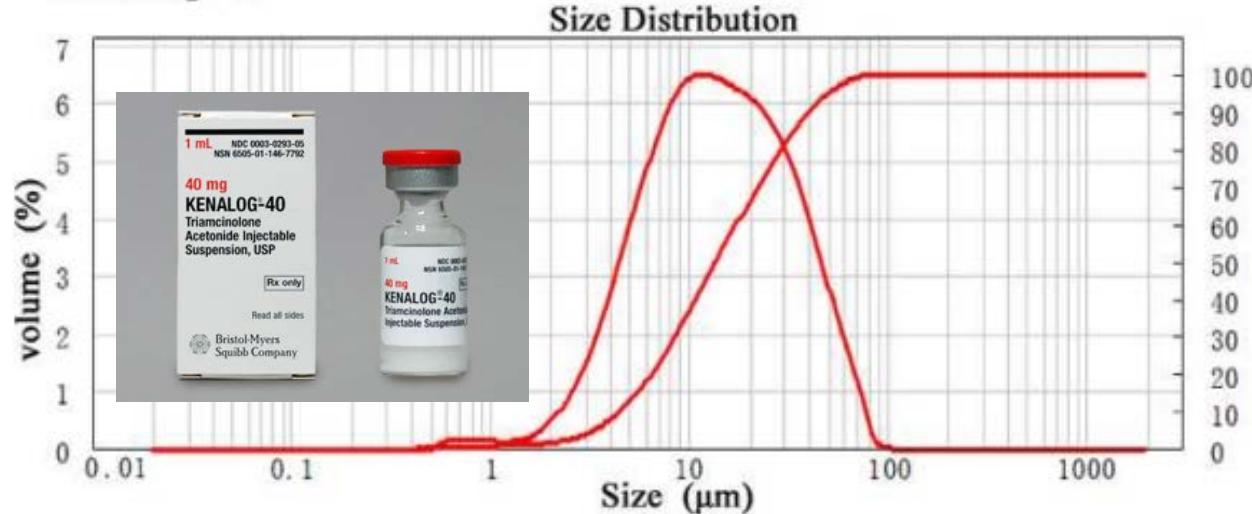
Develop models and recommendations to correlate *in vitro* and *in vivo* drug release profiles

# Kenalog40 vs Triesence

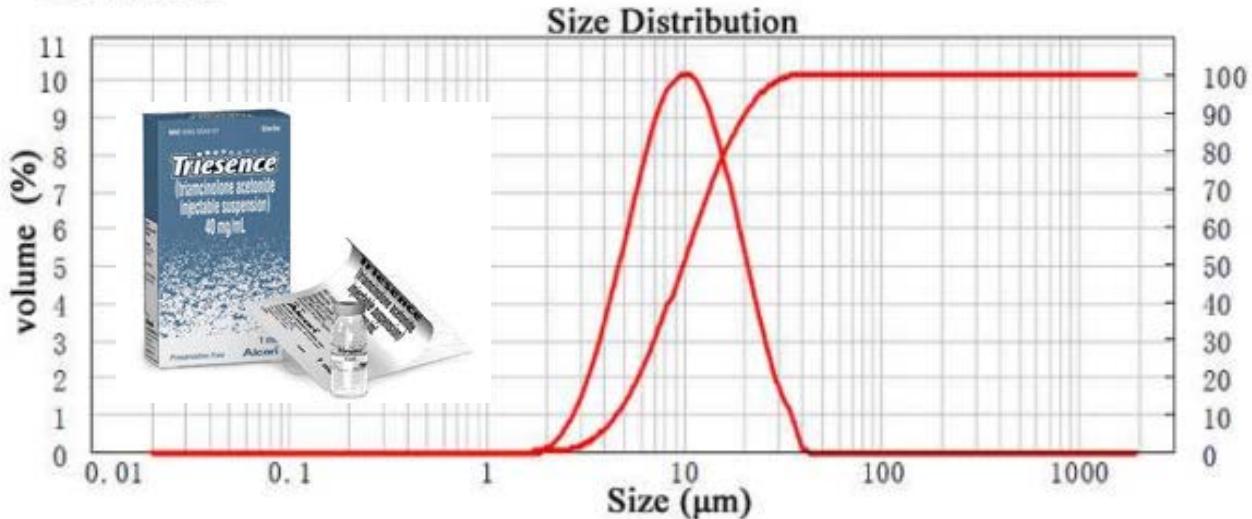
treat uveitis by intraocular injection

- API: Triamcinolone
- Crystalline
- Broad size distribution
- Benzyl alcohol preservative
- Salts, surfactants
- For intra-articular and intramuscular use
- Half-life in the eye (4mg, rabbit): 23d
- Used off-label for uveitis

Kenalog-40



Triesence



# Kenalog40 vs Triesence

*treat uveitis by intraocular injection*

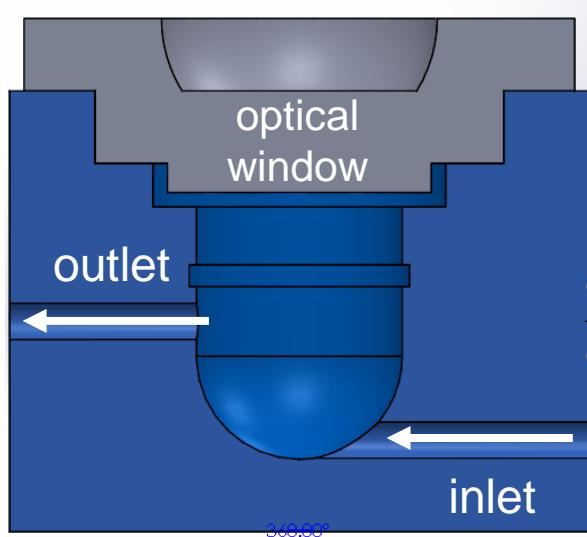
	Kenalog40	Triesence
<b>Triamcinolone Acetonide (Active Ingredient)</b>	40 mg/mL	40 mg/mL
<b>Sodium Chloride</b>	0.65%	Possibly Reduced (concentration not available)
<b>Benzyl Alcohol</b>	0.99% (w/v)	Not Included
<b>Carboxymethylcellulose Sodium</b>	0.75%	0.5%
<b>polysorbate 80</b>	0.04%	0.015%
<b>pH (adjusted with NaOH or HCl)</b>	5.0 – 7.5	6 – 7.5
<b>Additional Inactive Ingredients</b>	Water	Calcium chloride (dihydrate), magnesium chloride (hexahydrate), potassium chloride, sodium acetate (trihydrate), sodium citrate, and water

# Static vs Flow system

	Static	Flow
Plumbing	Single vial per sample	Requires separate pumps, tubing for each sample
Sampling	Discrete	Continuous
Analytics	Simple	Difficult (dilute analyte)
Control of dissolution rate	Based on sample interval Not representative of physiology	Readily adjusted

# In vitro flow cell design

Teflon flow cell



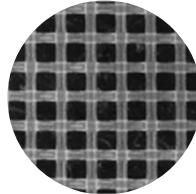
Lid with quartz window



Fittings and mesh filters



10um pores

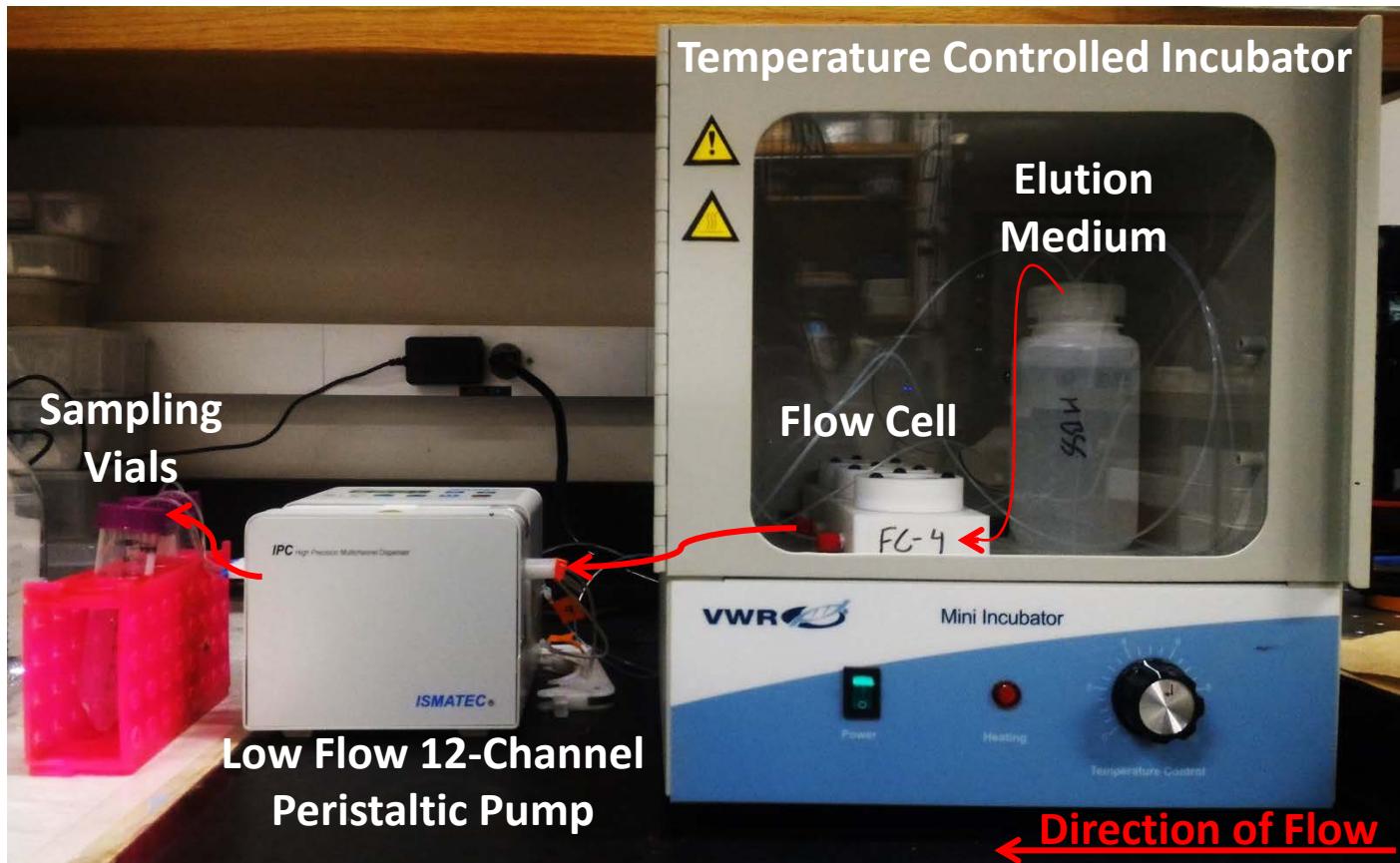


+



# In vitro system

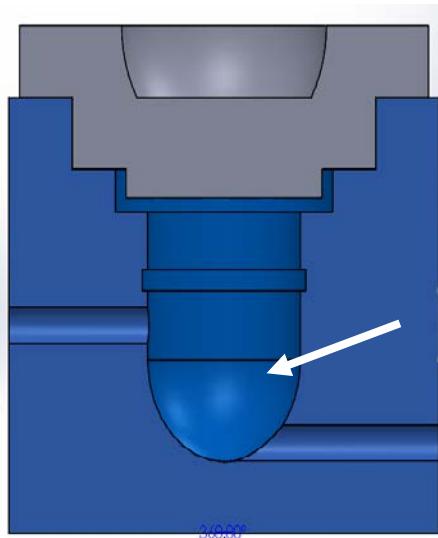
- Sample daily
- Flow rate
  - 1.5 mL/day
  - 7.5 mL/day
  - 15 mL/day
- Analysis
  - HPLC – UV-Vis
  - Extract ½ life



# The elution matrix

*simulating vitreous fluid*

- Volume
- Viscosity
- Chemical composition:
  - HBSS
  - Hyaluronic acid
  - Collagen



Concentration of Hyaluronic Acid (mg/mL)	Viscosity (cP)
0 (Hank's Balanced Salt Solution, HBSS)	1.0
0.05	$1.1 \pm 9 \times 10^{-3}$
0.1	$1.3 \pm 5 \times 10^{-3}$
0.5	$3.0 \pm 0.01$
1.0	$4.7 \pm 0.6$

1.5 mL  
(rabbit vitreous volume)

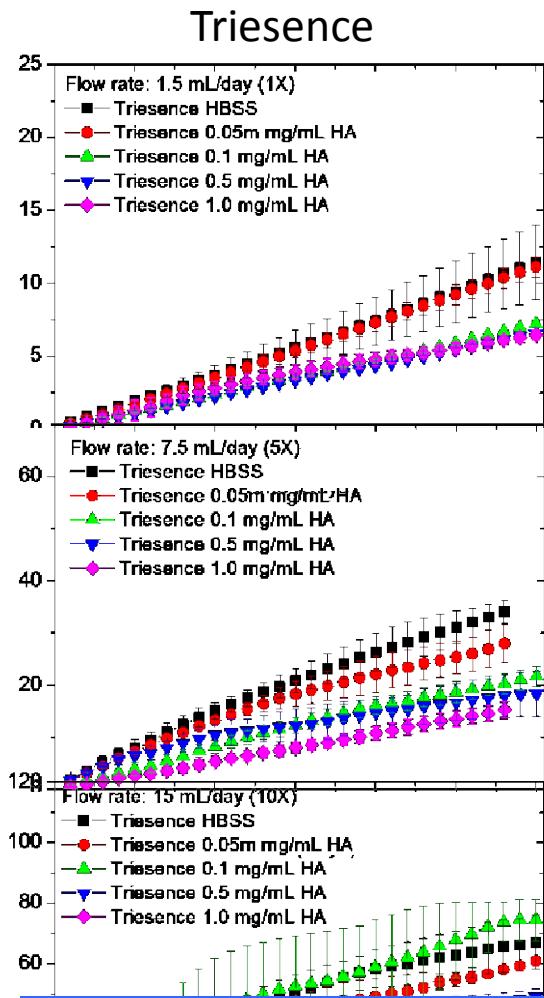
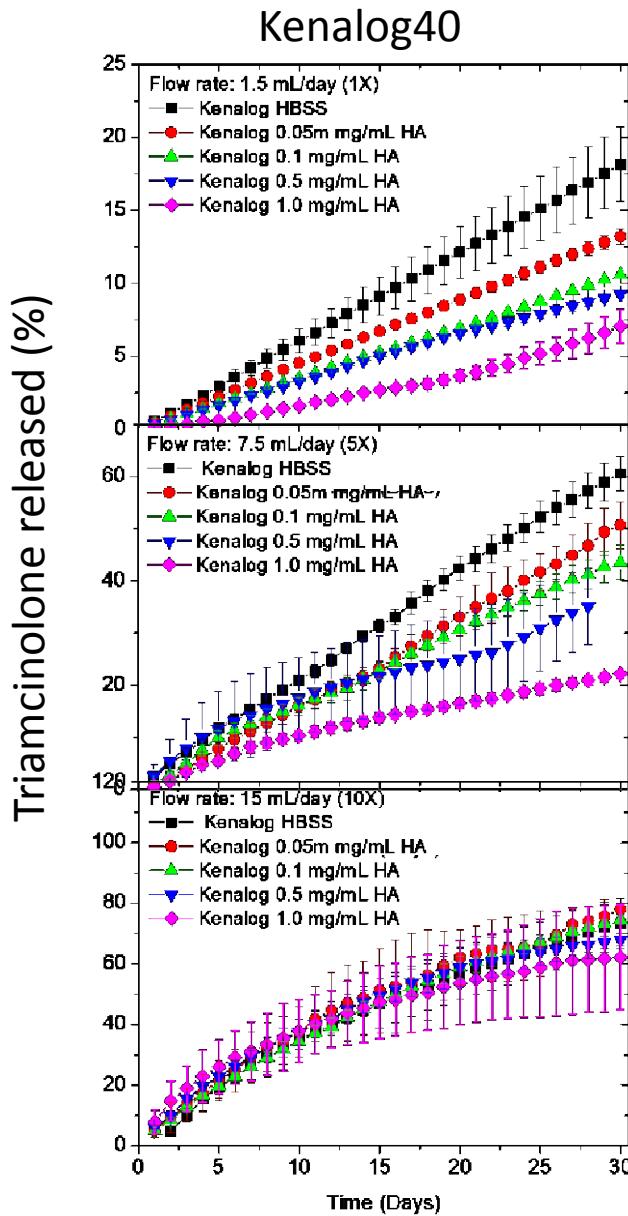
# Conditions tested

*for Kenalog40 and Triesence formulations*

Flow rate	Concentration of hyaluronic acid (ug/mL) Increasing viscosity -->				
	0	50	100	500	1000
static					
1 uL/min					
5 uL/min			Closest match to in vivo		
10 uL/min					

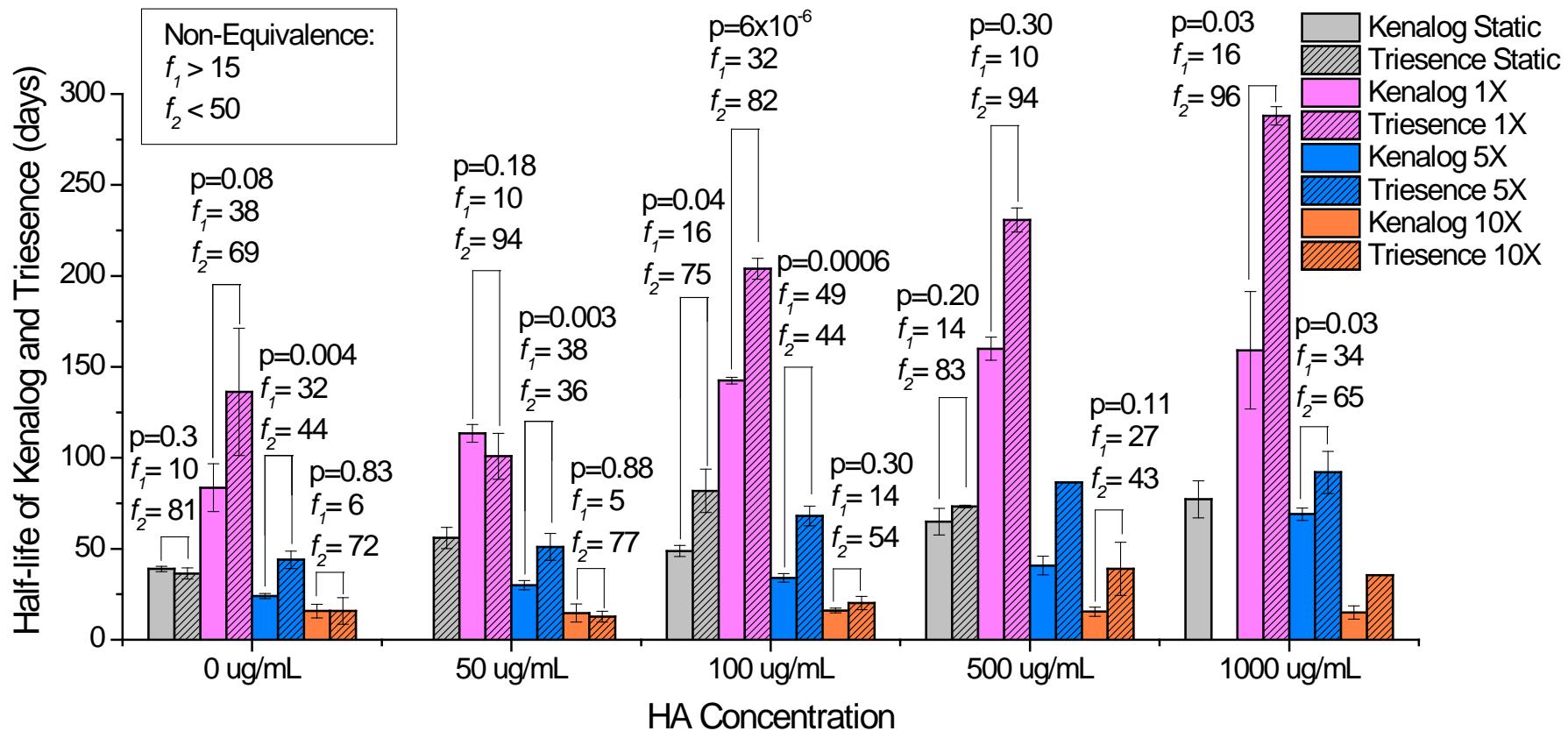
Injected dose: 4 mg • Volume of vitreous: 1.5 mL • static = 1 turnover/day

# In Vitro Results

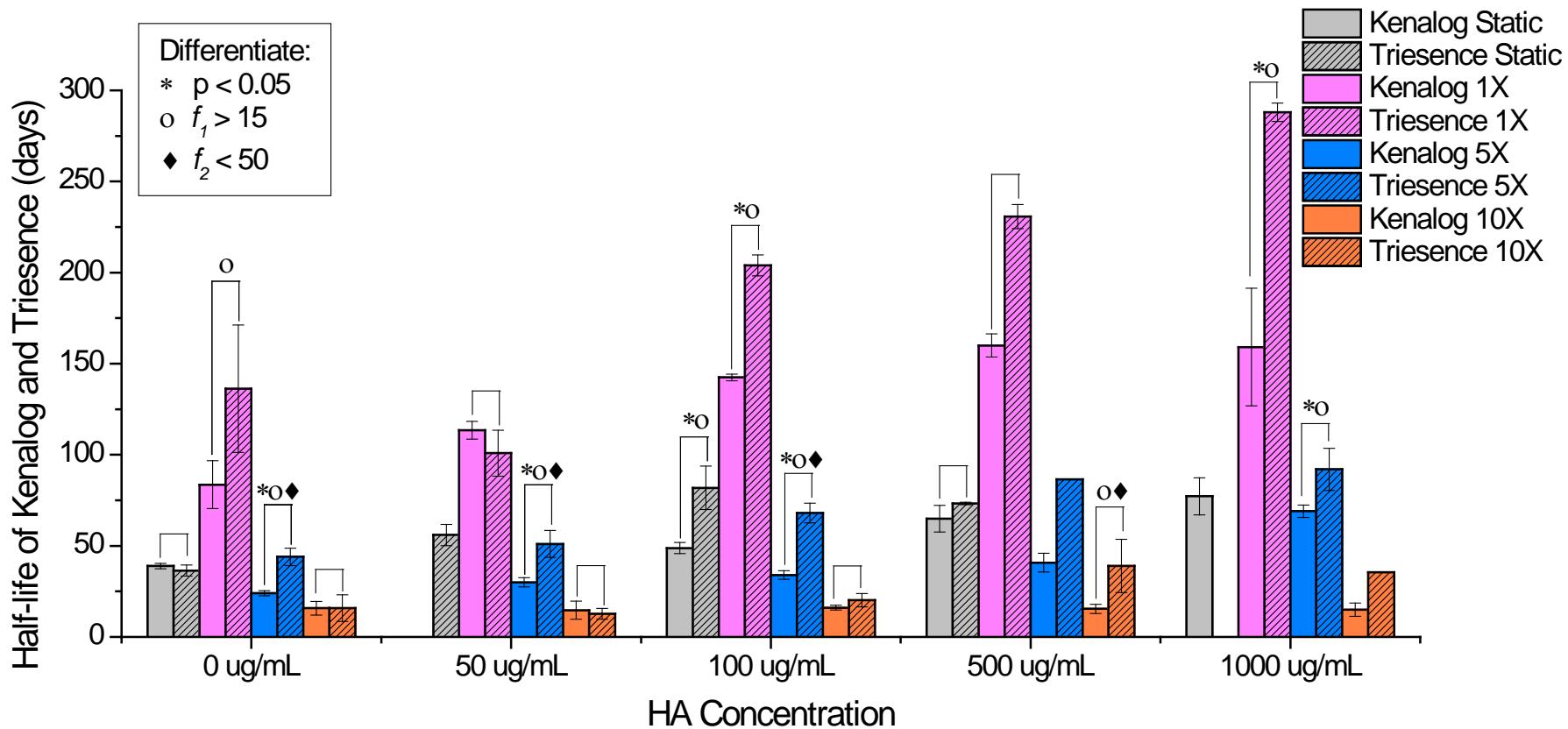


Drug clears faster at high flow  
Intermediate flow rates can give best discrimination  
Drug clears slower with increasing [HA]  
Triesence generally clears slower than Kenalog

# In Vitro Results



# In Vitro Results



# Conditions that discriminate

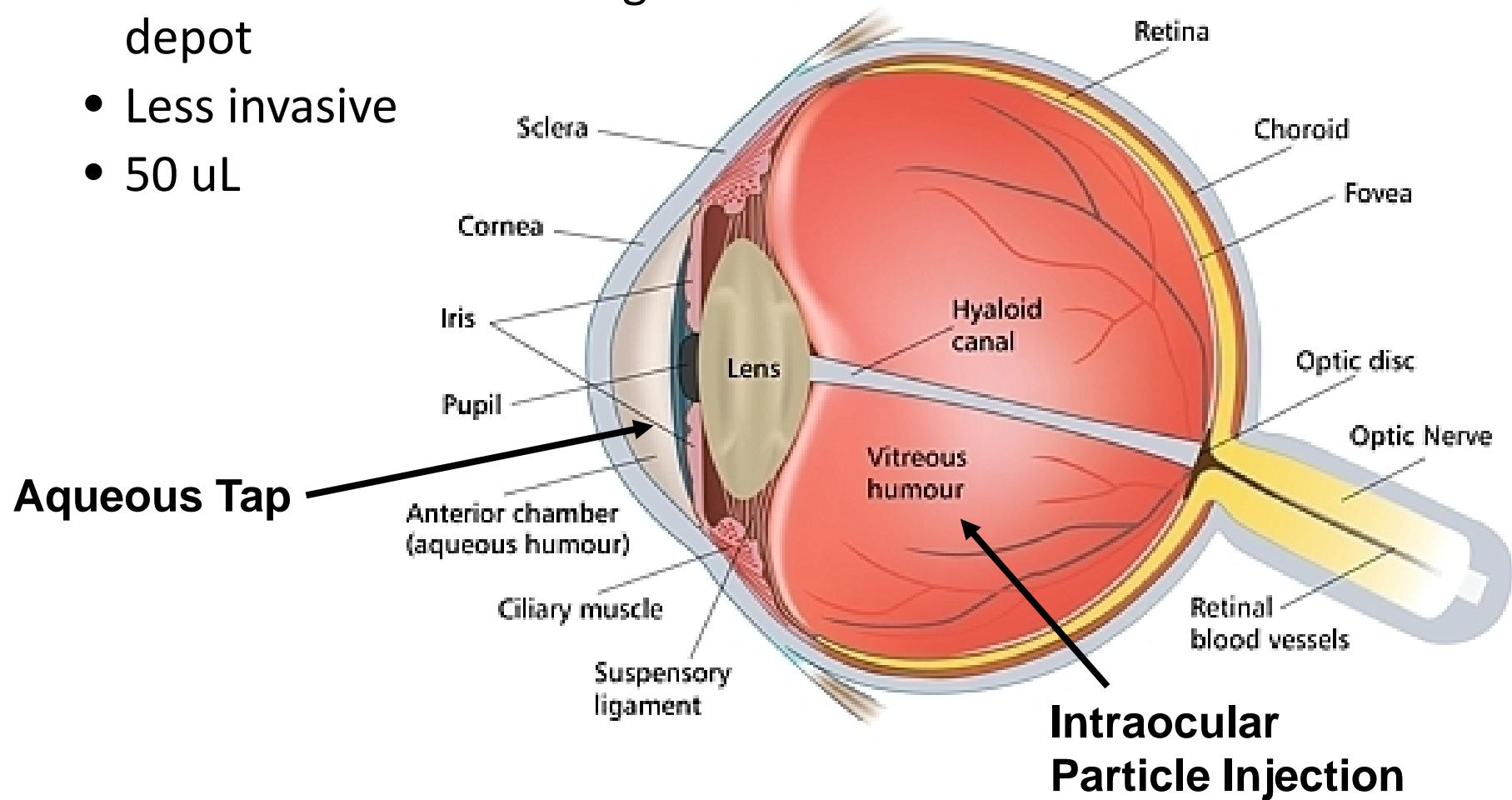
*Kenalog40 and Triesence formulations*

Flow rate	Concentration of hyaluronic acid (ug/mL) Increasing viscosity -->				
	0	50	100	500	1000
static	no	no	no	no	no
1 uL/min	no	no	no	no	no
5 uL/min	yes	yes	Closest match to in vivo	no	no
10 uL/min	no	no	no	yes	no

Injected dose: 4 mg • Volume of vitreous: 1.5 mL • static = 1 turnover/day  
“Yes” indicates  $p < 0.05$ ,  $f_1 > 15$ ,  $f_2 < 50$

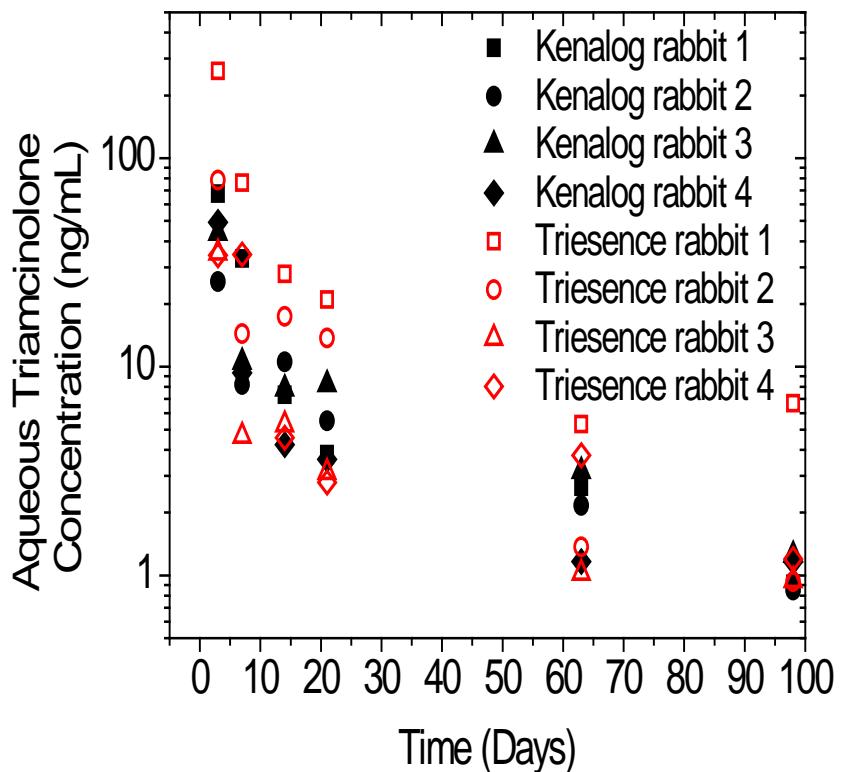
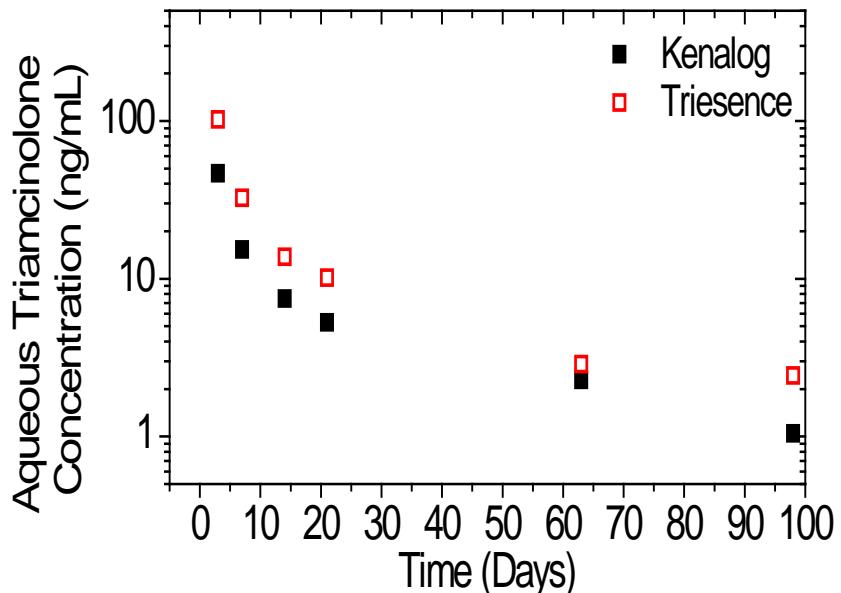
# In Vivo Results

- Tap aqueous
  - Prevent removal of drug depot
  - Less invasive
  - 50  $\mu\text{L}$



# In Vivo Results

- Tap aqueous
  - Prevent removal of drug depot
  - Less invasive
  - 50  $\mu$ L
- HPLC-MS analysis
- Toxicity: 1 TE rabbit
- Cannot differentiate
  - High variation between rabbits



# Summary and Recommendations

1. Continuous flow system is convenient in evaluating the pharmacokinetics (esp. particulate based systems) in vitro.
2. When comparing two similar formulations, specific flow rate and dissolution viscosity conditions can be adjusted to differentiate closely similar ophthalmic formulations:  
**Kenalog40 vs Triesence.**
3. Flow system can be adjusted to match *in vivo* performance.
4. Studies on the effect of inactive ingredients on pharmacokinetics are needed.

# *Thanks*

Joanna Wang  
Tushar Kumeria  
Yuan Zou (FDA)  
Yan Wang (FDA)



FDA Grant 1U01FD005173-01

