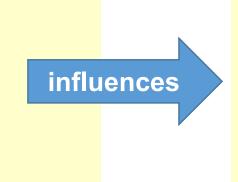


<u>Soumyarwit Manna<sup>1,2</sup>, Yong Wu<sup>2</sup>, Peter E Petrochenko<sup>1</sup>, Bonhye Koo<sup>1,2</sup>, Lynn Chen<sup>1</sup>,</u> Yixuan Dong<sup>3</sup>, Xiaoming Xu<sup>3</sup>, Stephanie Choi<sup>1</sup>, Darby Kozak<sup>1</sup>, Yan Wang<sup>1</sup>, Jiwen Zheng<sup>2</sup>

## BACKGROUND

- Bupivacaine (BPV), is an amino-amide local anesthetic, with a short half-life ( $t_{1/2} \sim 2.7$  hours)
- BPV, when encapsulated in a multivesicular liposome (MVL) formulation, exhibits *sustained release* characteristics (t<sub>1/2-BPV</sub> ~ 34 hours)
- Multivesicular liposomes (MVLs) consist of a non-lamellar honeycomb structure with *non-concentric* aqueous chambers
- Information is limited on the mechanism of drug release from MVLs required to advance MVL drug product development, quality assessment, and generic formulation equivalence
- Unique physicochemical properties and complex drug release characteristics *challenge* the design and development of appropriate in vitro release test (IVRT) methods
- It is *hypothesized* that altering IVRT conditions can affect the mechanism of drug release from the MVLs
- Speed and orientation of agitation
- Composition of release medium (presence of protein)
- ≻ pH
- > Temperature



- > Change in liposomal vesicles (e.g. size and morphology of the outer lipid membrane and inner vesicles)
- Drug (e.g. solubility, ionization, and partitioning)
- > Lipids (e.g. arrangement and phase transitions)

# OBJECTIVES

- Obtain a mechanistic understanding of the physical phenomena involved in drug release from MVLs
- Characterize the vesicle size and morphology of the BPV-MVL complex formulation
- Conduct IVRTs to evaluate the effect of different parameters such as a) agitation, b) composition of release medium, c) temperature, d) pH, and e) dissolution set-up on the release mechanism of the drug over a period of 7 days - expected complete drug release profile (5 times the t<sub>1/2-BPV</sub>)

# METHODS

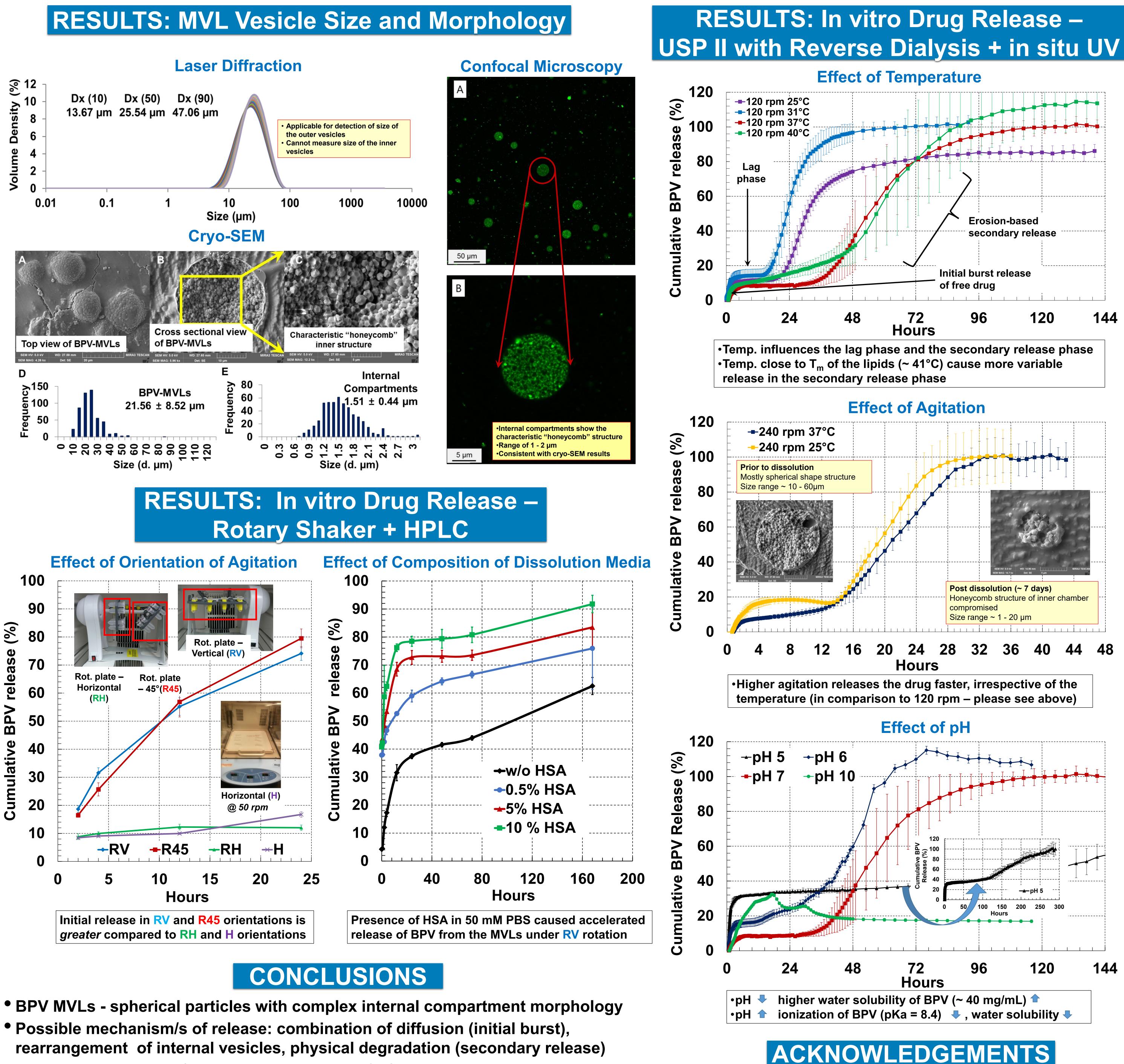
- MVL vesicle size and morphology: Cryo-scanning electron microscopy (Cryo-SEM), laser diffraction, confocal microscopy
- IVRT set-up and drug release detection techniques:
- $\succ$  Rotary shaker (Thermo Scientific<sup>TM</sup> Tube Revolver / Rotator), n = 3

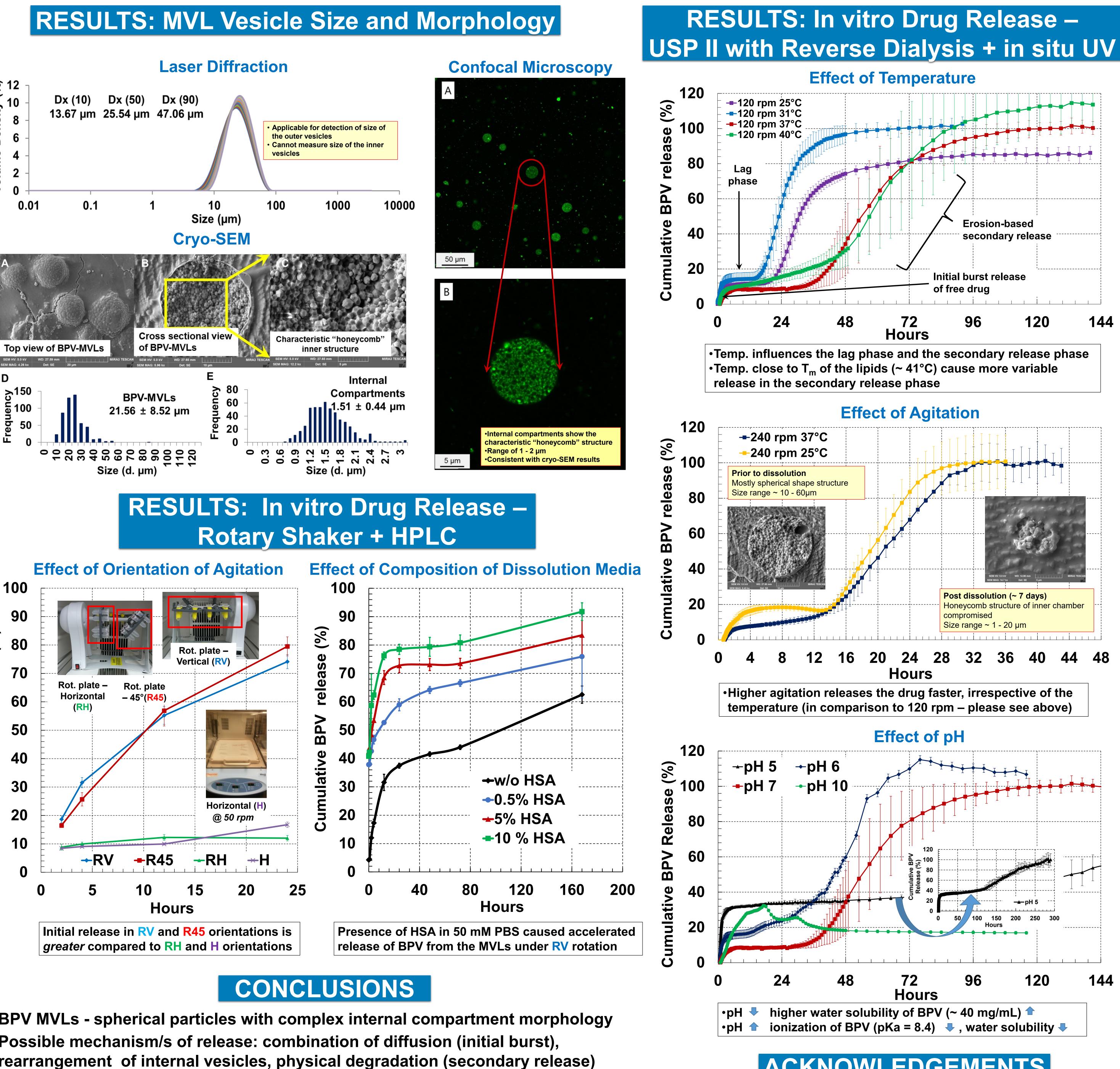
Dissolution media		50 n	) mM PBS (pH 7) with varied conc. of human serum albumin (HSA )						
Temp.	37°C	Rotation		12 rpm		Conc. BPV - MVLs	0.782 mg / mL		
HPLC Column			Agilent ZORBAX SB-CN 4.6 × 150 mm, 5µm						
Mobile phase			40% Acetonitrile + 0.01% TFA; pH 2.8						
Flow rate			1 mL / min		Column Temperature		30°C		
λ <sub>BPV</sub>			263 nm		Retention time		~ 2.1 min		

> USP II apparatus with Reverse dialysis (Teledyne Hanson Research), n = 3

Dissolution media	pH 5, pH 6 (citrate phosphate buffer) ; pH 7 (phosphate buffer); pH 10 (carbonate buffer)					
Temperature25°		C, 31°C, 37°C and 40°C	Agitation	120 rpm, 240		
Dialysis cartridges		100 kDa (MWCO) cellulose ester membrane conditioned wit				
Drug detection		In situ UV-Vis fiber optic probe				
Conc. BPV - MVLs		0.58 mg / mL				

- 0 rpm
- vith SDS





- IVRT– Rotary shaker (captures a *biphasic* profile) > Orientation of agitation and presence of HSA influence BPV release • IVRT– Reverse dialysis – USP II coupled with in-situ UV fiber optic (captures a
- *triphasic* profile)
- triggered by physical degradation of MVLs > Advantage - continuous monitoring of BPV release
- were impacted by the temperature, agitation and pH

# **Developing Physicochemical Characterization and In vitro Release Test Methods** to Probe Drug Release Mechanism from Multivesicular Liposomes

<sup>1.</sup> Office of Research and Standards, Center for Drug Evaluation and Research, FDA, Silver Spring, MD <sup>2.</sup> Office of Science and Engineering Laboratories, Center for Devices and Radiological Health, FDA, Silver Spring, MD <sup>3.</sup> Office of Testing and Research, Center for Drug Evaluation and Research, FDA, Silver Spring, MD

> Discern both the initial diffusion burst release (~ 2 – 4 hr) and the secondary release, likely

> Initial burst release was primarily influenced by the pH, while the lag and secondary release

The authors would like to thank the FDA Advanced Characterization Facility (ACF) and CDRH/Office of Science and Engineering laboratories/Division of Biology, Chemistry and Material Science for instrument use. Dr. Manna was supported in part by an appointment to the Oak Ridge Institute for Science and Education (ORISE) Research Participation Program at the Center for Drug Evaluation and Research administered by the ORISE through an agreement between the U.S. Department of Energy and FDA/CDER.

The views expressed in this poster do not necessarily reflect the official policies of the U.S. Food and Drug Administration; nor does any mention of trade names, commercial practices, or organization imply endorsement by the United States Government.

## DISCLAIMER